Ancient Monuments Laboratory Report 6/86

COPPER ALLOY WASTE FROM THE GREEN , NORTHAMPTON .

Justine Bayley

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Summary

Nine pieces of waste were analysed qualitatively by XRF and some were shown to be bell metal, including one piece adhering to a cope fragment of bell mould. Most of the material was late medieval (1350 - 1500) in date.

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The finds submitted for examination and analysis are listed below together with individual notes and comments. Most are late medieval (1350-1500) in date.

Finds No

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91
          Copper alloy waste; ? offcut from the end of a bar
          Copper alloy waste; spill of molten metal
95
          Copper alloy waste; spill of molten metal
100
          Copper alloy; ? corrosion products on lump of earth
102
119
          Copper alloy waste; ? spilt molten metal
          Copper alloy waste; droplet
121
          Copper alloy waste; ? droplets
129
          Copper impregnated ?? organic material
134
          Copper alloy fragments
138
143
          Copper alloy waste; ? spilt molten metal
          Copper alloy waste; metal adhering to a rim piece of
144
          a bell mould (cope). The metal has run between the two
          halves of the mould which cannot have been sufficiently
          well luted together.
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Most of the pieces were analysed qualitatively by energy dispersive X-ray fluorescence. The elements detected in each piece are listed below. Those elements appearing in brackets were not significant components of the alloy.

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91
          copper tin (lead antimony)
 95
          copper lead
100
          copper tin (zinc lead antimony)
          copper lead (tin zinc antimony)
119
          copper (lead zinc)
121
129
          copper tin (lead)
138
          copper tin (lead)
143
          copper tin (lead)
144
          copper tin (lead)
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Both its context and composition suggest 144 is bell metal, ie high tin bronze. Some of the other samples may also we waste metal from the same process. Those containing significant amounts of lead, 95 and 119, are not suitable for making bells. The zinc and antimony are only present in trace amounts, probably a percent at most in the metal.